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Chapter 21 Nuclear Chemistry

Name Unit 11: Nuclear Chemistry Review Packet Regents Chemistry 1. Base your answer to the following question on Given the nuclear equation: ${}_{11}^1\text{H} + X \rightarrow {}_{63}^{63}\text{Li} + {}_{2}^4\text{He}$ A) ${}_{94}^{94}\text{Li}$ B) ${}_{94}^{94}\text{Be}$ C) ${}_{105}^{105}\text{Be}$ D) ${}_{106}^{106}\text{C}$ The particle represented by X is A) ${}_{53}^{53}\text{Fe}$ B) ${}_{137}^{137}\text{Cs}$ C) ${}_{198}^{198}\text{Au}$ D) ${}_{220}^{220}\text{Fr}$ 2. Which isotope will spontaneously decay and

emit particles with a charge of +2? A)alpha B)beta C)neutron D)positron 3.Which of these ...

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2015040 Chapter 21 Review Nuclear Chemistry Section 4 Answers spontaneously, emitting energy, they are said to be radioactive. Nuclear chemistry is the study

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